Superoxide Radical Anion Oxidative Ionization in Negative LDTD-MS/MS Analysis of Serotonin

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OVERVIEW

Purpose

• Understanding of the ionization process for the formation of M-2 primary mass of Serotonin and application of this method for sample analysis.

Method

- Serum protein precipitation.
- Samples dried and analyzed by LDTD-MS/MS.

Quantification

- Linearity: r > 0.995 over the calibration range.
- Between-run accuracy, values between 105.6 and 113.0 are obtained and precision results are lower than 4.6% CV.
- Samples are analyzed with a runtime of 8 seconds using LDTD-MS/MS technique.

INTRODUCTION

Serotonin is an indoleamine neurotransmitter derived from the amino acid tryptophan. Its biological functions are complex, influencing various physiological processes such as mood regulation, memory, and even contributing to undesirable effects like vomiting. The measurement of serotonin concentration in serum is commonly used as a biomarker for diagnosing conditions such as carcinoid syndrome and other related disorders. During method development, the major ion observed is M-2 when using the Laser Diode Thermal Desorption (LDTD) ionization source.

To better understand the ionization process, several experiments are conducted. Ultimately, a reliable method was developed for analyzing serotonin levels in human serum.

LUXON Ionization Source:

The Luxon Ion Source (Figure 1) is the second-generation sample introduction and ionization source based on the LDTD technology for mass spectrometry. The Luxon Ion Source uses a Fiber-Coupled Laser Diode (Figure 2) to obtain unmatchable thermal uniformity giving more precision, accuracy and speed.

The process begins with dry samples which are rapidly evaporated using indirect heat. The thermally desorbed neutral molecules are carried into a corona discharge region. High-efficiency protonation and strong resistance to ionic suppression characterize this type of ionization and is the result of the absence of solvent and mobile phase. This thermal desorption process yields high-intensity molecular ion signal in less than 1 second sample-to-sample and allows working with very small volumes.





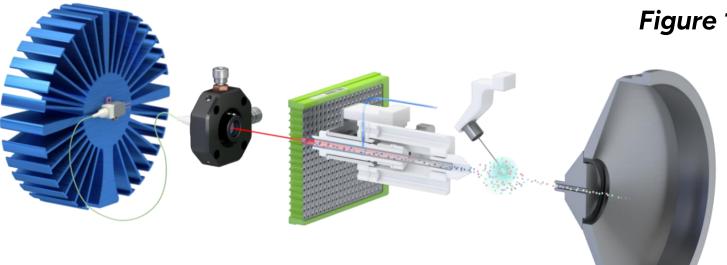
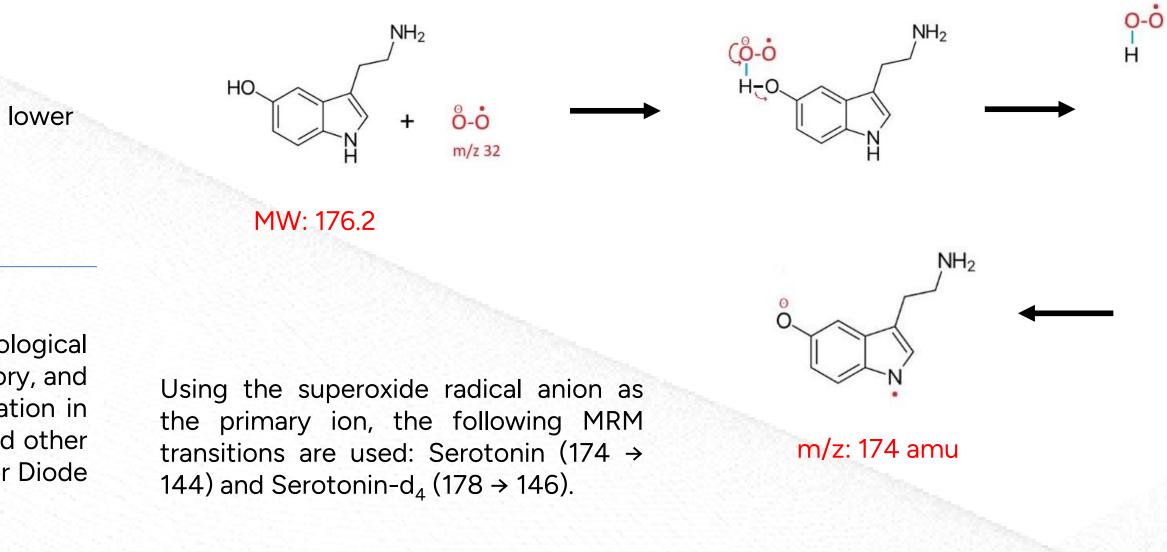


Figure 2 Schematic of the Luxon Ion Source

IONIZATION PROCESS

The first step in the oxidative ionization of serotonin by the superoxide radical anion involves the formation of the M+32 (-) species. A signal for this species is observed in the negative Q1 scan experiment for serotonin (208 amu) and serotonin-d₄ (212 amu). The M-H⁻ species are also observed, but the major ion signal corresponds to the M-2H⁻ species, with serotonin (174 amu) and serotonin-d₄ (178 amu) being the primary ions detected. Previous reports by Hassan et al. (2017)¹ suggest that this ionization process can be applied to serotonin analogs. Using the superoxide radical anion as the primary ion, the following MRM transitions are used: Serotonin (174 \rightarrow 144) and Serotonin-d₄ (178 \rightarrow 146). The following ionization process are suggested:



MFTHOD

Extraction

BSA solution (30 mg/mL) and Human serum were spiked to generate calibration curves and QC.

Automated SALLE extraction

100 µL of serum are transferred into a deep-well extraction plate on a vortexer system and mixed with 100 μ L of the internal standard solution (Serotonin-d₄, 100 ng/mL in water). Then, an extraction buffer (500 μ M K₂HPO₄ and 50 mM NaOH in saturated solution of NaCl) and 300 µL of acetonitrile were added, mixed, and centrifuged. Finally, 4 µL Butylated hydroxytoluene (BHT) at 1 mg/mL in acetonitrile, followed by 4 μ L of the upper-layer were spotted onto LazWell96 plates and evaporated to complete dryness before analysis by LDTD-MS/MS.

Instrumentation

- Ion source: Phytronix Luxon Ion Source S-960
- Mass spectrometer: Sciex, Q-Trap System 5500

Luxon Parameters

- Laser power pattern:
- Increase laser power to 45% in 3 s
- Carrier gas flow: 3 L/min (Air)

MS Parameters

- APCI (-)
- Curtain: 10
- CAD: 8 • Time: 50 msec
- MRM mode

Table 1 MRM transitions parameters

Compound	Q1 (m/z)	Q2 (m/z)	CE (eV)		
Serotonin	174.0	144.0	-25		
Serotonin-d ₄	178.0	146.0	-25		

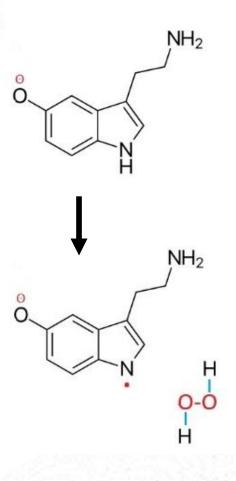




Figure 3 Azeo: Automated extraction system

RESULTS

Linearity

Table 2 shows the inter-run correlation coefficients for Serotonin. Values greater than 0.995 are obtained.

Precision and Accuracy

For the accuracy and precision evaluation, the following acceptance criteria are used:

- Each concentration must not exceed 15 %CV.
- Each concentration must be within 100 ± 15% of the nominal concentration.

For the inter-run precision and accuracy experiment, each QC is analyzed in triplicate, on five different runs. Table 3 shows the inter-run precision and accuracy results for Serotonin. The obtained %CV is below 15% and the accuracy is within 15% of the nominal value.

Stability

Wet stability of sample extracts:

Following the extraction, the sample extracts are kept at 4°C in closed containers protected from light. After 1 day, sample extracts are spotted on a LazWell™ plate, dried and analyzed. Precision and accuracy of QC samples are reported in Table 4. All the results are within the acceptable criteria range for 1 day at 4°C.

Dry stability of samples spotted on LazWell:

Extracted samples are spotted onto a LazWell™ plate, dried and kept at room temperature for 1 hour before analysis. The precision and accuracy results of QC				Serotonin	LC (ng/mL)	LUXON (ng/mL)	%Diff (%)				
samples are reported in Table 4. All the results are within the acceptable criteria						<u>M1</u>	148.1	157.0	-5.8%		
range for 1 hour at room temperature.					M2	71.9	82.2	-13.3%			
								M3	120.4	125.6	-4.2%
Table 4 Wet an	d dry sta	bility of Se	erotonin					M4	197.6	205.3	-3.8%
Devenetore						(140)		M5	119.6	138.1	-14.3%
Parameters	Dry Sta	bility (1 ho			ability (1 da			M6	169.3	180.4	-6.4%
QC	QCL	QCM	QCH	QCL	QCM	QCH		M7	49.2	55.6	-12.3%
Conc (ng/mL)	71.9	120.4	148.1	71.9	120.4	148.1		M8	80.4	85.2	-5.8%
Ν	3	3	3	3	3	3		M9	64.3	71.9	-11.1%
	022	135.4	168.4	71.7	121.5	149.6		M10	93.8	99.3	-5.7%
Mean (ng/mL)	82.3				121.5	149.0		M11	80.2	85.5	-6.4%
%CV	0.9	3.1	1.4	2.1	1.4	1.0		M12	107.1	117.1	-8.9%
%Nom	114.4	112.5	113.7	99.8	100.9	101.0					

CONCLUSION

- Ionization process via Superoxide radial anion are obtained.
- Efficient Automated Salt Assisted Liquid-Liquid Extraction (SALLE) is used to extract serotonin
- High-throughput analysis using LDTD-MS/MS
- Linearity, accuracy, precision and stability within the acceptance criteria.
- Sample-to-sample analysis of 8 seconds

REFERENCES

1) Hassan I et Al. (2017). Oxidative Ionization under certain negative-ion mass spectrometric conditions, J. Am. Soc. Mass Spectrom., 28:270-277, https://doi.org/10.1007/s13361-016-1527-5.

COI Disclosure: I have a financial relationship with Phytronix Technologies as a salaried employee.

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Table 2 Inter-run calibration curve correlation coefficients (r)

MP-332

	Serotonin
Curve 1	0.99991
Curve 2	0.99988
Curve 3	0.99969
Curve 4	0.99917
Curve 5	0.99973

Table 3 Inter-run precision and accuracy

QC-L	QC-M	QC-H
71.9	120.4	148.1
15	15	15
81.3	127.2	162.3
4.3	2.3	4.6
113.0	105.6	109.6
	71.9 15 81.3 4.3	71.9120.4151581.3127.24.32.3

Cross validation study

Real patients' serum samples (N=12) have been tested with this method to correlate with results obtained by traditional LC-MS/MS. The percentage of difference between the values are evaluated. A difference below 15% is obtained. Results are reported in Table 5.

 Table 5 Cross validation results